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Can Carbomer be effectively replaced with natural polymers?

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Introduction and aim

- Demand for 'greener' alternatives to synthetic polymers
- Natural rheology modifiers great variety of structures; inferior to synthetic polymers in the formulation stability and sensory aspects
- Classification from Food science into thickening and gelling agents

The aim of this project:

To replace Carbomer using a combination of natural polymers of polysaccharide type, focusing on the rheological, textural and sensory properties

Materials: natural polymers used

'Thickening' polysaccharides

Xanthan gum

Cellulose gum

Konjac gum

Guar gum

Gelling' polysaccharides

Carrageenan

Gellan gum

Thickening mechanism

Main factors: Temperature Molecular mass and shape



Gelling mechanism



Main factors: Temperature Presence of ions

Materials: emulsion formulation

1. Each natural polymer was first assessed as:

- a single polymer in hydrogel
- a single polymer in emulsion

- 2. This was followed by:
 - a combination of polymers in hydrogel
 - a combination of polymers in emulsion

	INCI Name	% (w/w)		
Phase A	Deionized water	Up to 100.00		
	Glycerin	2.0		
	Rheology modifier	Fit within the viscosity range		
Phase B	Butyrospermum Parkii (Shea) Butter	7.5		
	Capryl/ Caprylic Acid Triglyceride	7.5		
	Cetearyl Glucoside (and) Cetearyl Alcohol	4.0		
Phase C	Benzoic acid	0.5		
	Sodium hydroxide	As required to achieve target pH		
	Citric acid	As required to achieve target pH		
	Deionized water	q.s		

Methods: physico-chemical tests

Rheological tests

Oscillatory method

- <u>Oscillatory stress sweep</u> from 1 to 500 Pa, at 1 Hz (measures viscoelasticity)
- Parameters: complex modulus G* (rigidity) and phase angle δ (elasticity)

Continuous flow methods

- Shear rate sweep from 10 250 s⁻¹ (measures viscosity)
- Three-step thixotropy: 60 sec at 10 s⁻¹, 60 sec at 250 s⁻¹ and again 60 sec at 10 s⁻¹ (measures the rate of instant structure recovery)

Texture analysis

 Immersion/de-immersion of a cylinder probe (measures hardness, compressibility, stringiness and adhesiveness)





Methods: sensory tests

Part I: Sensory profiling test

Panel: 8 semi-trained panellists

<u>Objective</u>: To select the representative sample that is perceived to be the most similar to Carbomer.

<u>Assessment</u>: Four creams (including one with Carbomer), controlled conditions, six sensory attributes

Part II: Paired difference test

Panel: 40 naïve panellists (16 sighted, 32 blind-folded)

<u>Objective</u>: To determine if there is a perceived difference between the representative sample and Carbomer.

<u>Assessmen</u>t: Two creams, controlled conditions, assessed in parallel on different hands; **preference test** also performed.



Combinations of polymers tested, based on preliminary experiments



Code	Combination	
CG	Carrageenan – 1% Gellan gum HA – 0.2%	
CGG	Carrageenan – 0.6% Gellan gum HA* – 0.3% Gellan gum LA** – 0.2%	
GX	Guar gum 0.8% Xanthan gum – 1.5%	
СХ	Carrageenan – 1% Xanthan gum – 0.7%	

* HA – high acyl content **LA – low acyl content

The four combinations were chosen based on the rheological tests of more than 20 combinations.



Oscillatory stress sweep results for the polymer combinations

Continuous flow results



Texture analysis of the emulsions with different polymer combinations







Sample GX selected as the best match from the TA

Attribute	Mean rankings				7 \/	
	Carbomer	CG	CGG	GX	Z-Value	p-value
Absorbance	2.00	<u>2.43</u>	3.14	<u>2.43</u>	4.05	0.256
Spreadability	2.75	<u>3.00</u>	2.38	1.88	3.45	0.327
Wetness	3.21	2.43	<u>2.57</u>	1.79	5.89	0.117
Firmness	3.19	2.19	1.88	<u>2.75</u>	4.98	0.174
Integrity	3.50	2.00	1.50	<u>3.00</u>	12.00	0.007
Stickiness	2.00	<u>2.25</u>	2.88	2.88	3.75	0.290

Sample CG selected as the best match by the panel

Sensory test results, comparing emulsions with CG and Carbomer

Paired difference test

Preference test





Change in preference (blinded panel preferred emulsion CG over Carbomer), due to the white residue temporarily left on skin by CG

77% of participants perceived a difference

Conclusions

- Viscoelastic plot has been proven to be a useful rheological tool.
- Carbomer cannot be replaced by a single natural polymer.
- Natural 'gelling' agent(s) should be combined with 'thickening' agent(s) to increase the mechanical strength of the network.
- Sensory testing can not be completely replaced by instruments; the two methods should be combined.
- Linear/elastic polysaccharides could be included in the combinations, to remove the white residue and increase consumer acceptance.

References

Lukic, M., Jaksic, I., Krstonosic, V., Cekic, N. and Savic, S., A combined approach in characterization of an effective w/o hand cream: the influence of emollient on textural, sensorial and in vivo skin performance, International Journal of Cosmetic Science, 34(2), pp. 140-149 (2011)

Madlin, B., Baltazar, D., Stevic, M. C. and Tamburic, S., Water resistance vs. rheology control: an exploration, Cosmetics & Toiletries, Vol. 135, 2, 44-56 (2020)

Saha, D. and Bhattacharya, S., Hydrocolloids as thickening and gelling agents in food: a critical review, Journal of Food Science and Technology, 47(6), pp. 587-597 (2010)

Tai, A., Blanchini, R. and Jachowicz, J., Texture analysis of cosmetic/pharmaceutical raw materials and formulations, International Journal of Cosmetic Science, 36 291-304 (2014)

Tamburic, S., Craig, D.Q.M., Rheological Evaluation of Polyacrylic Acid Hydrogels, Journal of Pharmacy and Pharmacology 1(3), pp. 107-109 (1995)

Tamburic, S., Sisson, H., Cunningham, N. and Stevic, M.C., Rheological and texture analysis methods for quantifying yield value and level of thixotropy, SOFW Journal, 143(6), pp. 24-30 (2017)

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